

**SYNTHESIS OF 1-ARYL – 5 - TETRA – O – ACETYL – B – GLUCOSYL – 2 - S – BENZYL – 2, ISOTHIABIURETS**

**Ashish G. Sarap**

**Shri R. L. T. College of Science, Akola-444 001**  
**Maharashtra, India.**

**E-mail: [sarap.aashish1@gmail.com](mailto:sarap.aashish1@gmail.com)**

**ABSTRACT:**

Several 1-aryl – 5 - tetra – O – acetyl –  $\beta$  – glucosyl – 2 - S – benzyl – 2, isothiabiurets have been prepared by the interaction of aryl – S – benzyl isothiocarbamides and tetra – O – acetyl -  $\beta$  – glucosyl isocyanate. The structures of the newly synthesized compounds have been established on the basis of usual chemical transformations and IR, NMR and Mass spectral studies.

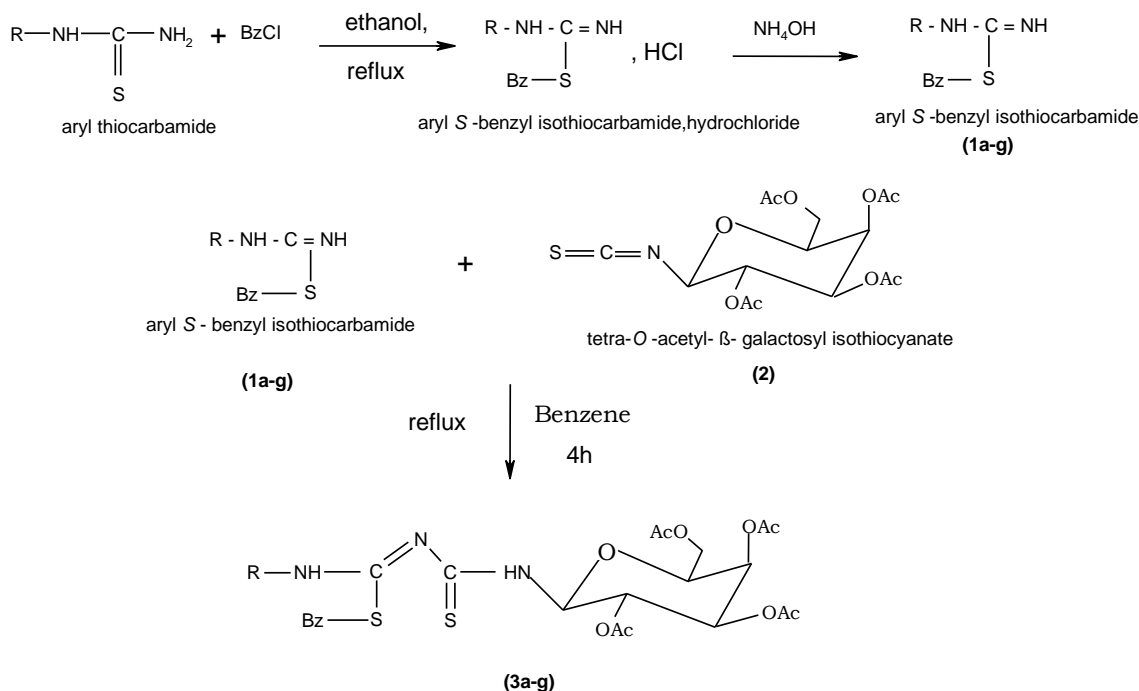
**Keywords:** *N* - glucosides, aryl – S – benzyl isothiocarbamides, glucosyl isocyanate, isothiabiurets.

**INTRODUCTION:**

Glycosyl Carbamides and Glycosyl thiocarbamides have great pharmacological and biochemical aspects<sup>1</sup>. Many of these derivatives have been found to possess wide applications in industry as Carbohydrate based detergent<sup>2</sup> and in medicine as anticancer agent<sup>3</sup> and antifungal agents<sup>4</sup>.

Glucosyl isocyanate and isothiocyanate<sup>5</sup>, likewise other glycosyl isocyanates has a significant role in synthetic Carbohydrate Chemistry. Many compounds have been synthesized by the use of glycosyl isocyanate, for example, galactosyl isocyanide<sup>6</sup>, galactosyl amino derivatives<sup>7</sup> and other heterocycles<sup>8</sup>. But there is no report on the synthesis of isothiabiurets having a  $\beta$  - glucosyl substituent. On the basis of knowledge gained on the work done on *N*-glucosylated isomonothio and dithiabiurets<sup>9-10</sup>, it was quite interesting to synthesize some new *N*-glucosylated thioamides.

In the present Communication, we report the synthesis of 1-aryl – 5 – tetra – O – acetyl –  $\beta$  – glucosyl -2-S-benzyl- 2, isothiabiurets by the interaction of tetra – O – acetyl –  $\beta$  - glucosyl isocyanate and aryl-S-benzyl isothiocarbamides. All the products obtained have been crystallized from alcohol. IR spectra of the products show characteristic absorption of prominent peaks, <sup>1</sup>H NMR spectra shows characteristic peaks due to N-H, acetyl and glycosidic protons, while the ESI Mass spectra shows peaks due to acetogalactose unit<sup>11-13</sup>.



### Scheme -1

Where, R = a) phenyl, b) *o*-Cl-phenyl, c) *m*-Cl-phenyl, d) *p*-Cl-phenyl, e) *o*-tolyl, f) *m*-tolyl, g) *p*-tolyl. Ac = -COCH<sub>3</sub> Bz = -CH<sub>2</sub>C<sub>6</sub>H<sub>5</sub>.

## RESULT AND DISCUSSION:

Several 1-aryl - 5 - tetra - O - acetyl - β - glucosyl -2-S-benzyl- 2, isothiobiurets (**3a-g**)(Sceme-1) were prepared by the condensation of aryl-S-benzyl isothiobiurets (**1a-g**) and tetra-O-acetyl- β-glucosyl isocyanate (**2**) in benzene for 4h. The reaction was monitored by TLC, after complete reaction, the solvent was distilled off and the resultant sticky residue was triturated with petroleum ether (bp 60-80<sup>0</sup> C) to afford the products (**3a-g**). All the products were crystallized from alcohol. Structures of all products synthesized were established on the basis of usual Chemical transformations and IR, NMR and Mass spectral studies.

## METHODOLOGY:

### 1. General methods:

Melting points are uncorrected. Optical rotations [ $\alpha$ ]<sub>D</sub> were measured on a Equip-Tronics digital polarimeter model no. EQ 800 in CHCl<sub>3</sub> at 39<sup>0</sup>C. IR spectra were recorded on a Perkin-Elmer spectrum RXI (4000-450cm<sup>-1</sup>) FTIR spectrometer. <sup>1</sup>HNMR were obtained on a Bruker DRX-300 (300 MHz FT NMR) NMR spectrometer for a sample in CDCl<sub>3</sub> solution with

TMS as an internal reference. The mass spectra were recorded on a Micromass Quattro II triple quadrupole mass spectrometer.

## **2. Aryl-S-benzyl isothiocarbamides (1a-g):**

The required aryl *S*-benzyl isothiocarbamides (**1a-g**) were prepared by the already known method<sup>14</sup>, which involves interaction of benzyl Chloride with aryl thiocarbamides in ethanol, refluxed for 90 min., followed by the basification of the resultant solution using dilute ammonium hydroxide to give the isothiocarbamides (**1a-g**).

## **3. Tetra-O-acetyl-β- glucosyl isocyanate (2):**

Tetra-*O*-acetyl- β - glucosyl isocyanate (**2**) was prepared by employing the classical Fischer's method<sup>15</sup> using lead cyanate instead of silver cyanate in anhydrous Xylene medium.

## **4. 1-aryl – 5 – tetra – O – acetyl – β – glucosyl -2-S-benzyl- 2, isothiobiurets (3a-g) (Scheme-1) .**

Condensation of aryl – *S* - benzyl isothiocarbamides (**1a-g**, 0.005M) and tetra-*O*-acetyl- β- glucosyl isocyanate (**2**, 0.005M, 2g), in benzene(20mL) was carried out on boiling water bath for 4h.. The reaction was monitored by TLC, after complete reaction, the solvent was distilled off and the resultant sticky residue was triturated with petroleum ether (bp 60-80<sup>0</sup> C) to afford the products (**3a-g**). All the products were crystallized from alcohol before recording the physical data (**table-1**). The purity of compounds (**3a-g**) was checked by TLC. The compounds formed have β configuration which is elaborated by the high value of vicinal coupling constant of proton at anomeric position<sup>16-17</sup>.

## **DATA ANALYSIS :-**

### **4.1 1-phenyl-5-tetra-O-acetyl - β - glucosyl -2-S- benzyl -2, isothiobiuret (3a).**

IR (KBr) :- ν 3336 (N-H); 2929 (Ar-H); 1749 (C=O); 1551 (C = N) 1366 (C-N); 1223 (C-O); 1044 (C=S). <sup>1</sup>HNMR (CDCl<sub>3</sub>) : δ 7.43-7.23 (m,11H,N-H(Ar.) and Ar-H); 6.97 (d, 1H,j=9Hz, NH(gal.)); 5.72 (t, 1H, j=9Hz, H<sub>1</sub>); 5.49 (d, 1H,j=3Hz, H<sub>4</sub>); 5.43-5.14 (m,2H, H<sub>2</sub>, H<sub>3</sub>); 4.35-3.95 (m, 5H, H<sub>5,6,6'</sub> and Ar-CH<sub>2</sub>); 2.17-1.60 (m, 12H, 4COCH<sub>3</sub>).

ESI Mass (M/z) : 632.4 (M<sup>+</sup>.+1), 331.3, 211.1, 169, 108.9, 90.9.

Anal. Calcd. for:- C<sub>29</sub>H<sub>33</sub>N<sub>3</sub>O<sub>9</sub>S<sub>2</sub>: C, 55.15; H, 5.22; N, 6.65; S,10.14, found, C, 54.99; H, 5.16; N, 6.51; S,10.01, %.

### **4.2 1-o-Cl-phenyl-5-tetra-O-acetyl - β - glucosyl -2-S- benzyl -2, isothiobiuret (3b).**

$^1\text{H}$ NMR ( $\text{CDCl}_3$ ) :  $\delta$  7.43-7.22 (m,10H,N-H(Ar.) and Ar-H); 6.38 (d, 1H,j=9Hz, NH(gal.)); 5.71 (t, 1H, j=9Hz,  $\text{H}_1$ ); 5.50 (d, 1H,j=3Hz,  $\text{H}_4$ ); 5.43-5.10 (m,2H,  $\text{H}_2$ ,  $\text{H}_3$ ); 4.35-4.01 (m, 5H,  $\text{H}_{5,6,6'}$  and Ar- $\text{CH}_2$ );2.16-1.97 (m,12H,4COCH<sub>3</sub>).Anal. Calcd. for:-  $\text{C}_{29}\text{H}_{32}\text{N}_3\text{O}_9\text{S}_2\text{Cl}$ : , 52.28; H, 4.84; N, 6.31; S, 9.62, found, C, 52.33; H, 4.78; N, 6.13; S, 9.52, %.

**4.3 1-*m*-Cl-phenyl-5-tetra-*O*-acetyl -  $\beta$  - glucosyl -2-*S*- benzyl -2, isothiobiuret (3c).**

IR (KBr) :-  $\nu$  3336 (N-H); 2930 (Ar-H); 1751 (C=O); 1548 (C = N) 1366 (C-N); 1223 (C-O); 1045 (C=S).  $^1\text{H}$ NMR ( $\text{CDCl}_3$ ) :  $\delta$  7.38-7.16 (m,10H,N-H(Ar.) and Ar-H); 6.38 (d, 1H,j=9Hz, NH(gal.)); 5.69 (t, 1H, j=9Hz,  $\text{H}_1$ ); 5.49 (d, 1H,j=3Hz,  $\text{H}_4$ ); 5.43-5.10 (m,2H,  $\text{H}_2$ ,  $\text{H}_3$ ); 4.32-4.00 (m, 5H,  $\text{H}_{5,6,6'}$  and Ar- $\text{CH}_2$ ); 2.16-1.67 (m, 12H, 4COCH<sub>3</sub>).

ESI Mass (M/z) : 666.4 ( $\text{M}^+$ ), 331.3, 211.1, 169, 108.9, 90.9(100%).

Anal. Calcd. for:-  $\text{C}_{29}\text{H}_{32}\text{N}_3\text{O}_9\text{S}_2\text{Cl}$ : C, 52.28; H, 4.84; N, 6.31; S,9.62, found, C, 52.22; H, 4.88; N, 6.22; S,9.69,%.

**4.4 1-*p*- Cl-phenyl-5-tetra-*O*-acetyl -  $\beta$  - glucosyl -2-*S*- benzyl -2, 4 isothiobiuret (3d).**

$^1\text{H}$ NMR ( $\text{CDCl}_3$ ) :  $\delta$  7.43-7.19 (m,10H,N-H(Ar.) and Ar-H); 6.38 (d, 1H,j=9Hz, NH(gal.)); 5.70 (t, 1H, j=9Hz,  $\text{H}_1$ ); 5.50 (d, 1H,j=3Hz,  $\text{H}_4$ ); 5.44-5.17 (m,2H,  $\text{H}_2$ ,  $\text{H}_3$ ); 4.35-3.99 (m, 5H,  $\text{H}_{5,6,6'}$  and Ar- $\text{CH}_2$ ); 2.16-1.68 (m, 12H, 4COCH<sub>3</sub>). Anal. Calcd. for:-  $\text{C}_{29}\text{H}_{32}\text{N}_3\text{O}_9\text{S}_2\text{Cl}$ : C, 52.28; H, 4.84; N, 6.31; S,9.62, found, C, 52.32; H, 4.79; N, 6.40; S,9.55,%.

**4.5 1-*o*-tolyl-5-tetra-*O*-acetyl -  $\beta$  - glucosyl -2-*S*- benzyl -2, isothiobiuret (3e).**

$^1\text{H}$ NMR ( $\text{CDCl}_3$ ) :  $\delta$  7.34-7.19 (m,10H,N-H(Ar.) and Ar-H); 6.97 (d, 1H,j=9Hz, NH(gal.)); 5.72 (t, 1H, j=9Hz,  $\text{H}_1$ ); 5.49 (d, 1H,j=3Hz,  $\text{H}_4$ ); 5.40-5.11 (m,2H,  $\text{H}_2$ ,  $\text{H}_3$ ); 4.30-4.01 (m, 5H,  $\text{H}_{5,6,6'}$  and Ar-  $\text{CH}_2$ );2.36(m,3H,Ar- $\text{CH}_3$ ); 2.17-1.60 (m, 12H, 4COCH<sub>3</sub>).Anal. Calcd. for:-  $\text{C}_{30}\text{H}_{35}\text{N}_3\text{O}_9\text{S}_2$ : C, 55.80; H, 5.42; N, 6.51; S,9.92 found, C, 55.84; H, 5.36; N, 6.42; S,9.86%.

**4.6 1-*m*-tolyl-5-tetra-*O*-acetyl -  $\beta$  - glucosyl -2-*S*- benzyl -2, isothiobiuret (3f).**

IR (KBr) :-  $\nu$  3333 (N-H); 2929 (Ar-H); 1749 (C=O); 1550 (C = N); 1365 (C-N); 1221 (C-O); 1046 (C=S).  $^1\text{H}$ NMR ( $\text{CDCl}_3$ ) :  $\delta$  7.36-7.12 (m,10H,N-H(Ar.) and Ar-H); 6.96 (d, 1H,j=9Hz, NH(gal.)); 5.72 (t, 1H, j=9Hz,  $\text{H}_1$ ); 5.48 (d, 1H,j=3Hz,

H<sub>4</sub>); 5.35-5.13 (m, 2H, H<sub>2</sub>, H<sub>3</sub>); 4.30-4.00 (m, 5H, H<sub>5,6,6'</sub> and Ar-CH<sub>2</sub>); 2.36 (m, 3H, Ar-CH<sub>3</sub>); 2.16-1.62 (m, 12H, 4COCH<sub>3</sub>). ESI Mass (M/z) : 646.5 (M<sup>+</sup>.+1)(100%), 331.3, 211.1, 169, 108.9, 90.9. Anal. Calcd. for:- C<sub>30</sub>H<sub>35</sub>N<sub>3</sub>O<sub>9</sub>S<sub>2</sub>: C, 55.80; H, 5.42; N, 6.51; S, 9.92, found, C, 55.84; H, 5.39; N, 6.59; S, 10.01%.

**4.7 1-p-tolyl-5-tetra-O-acetyl - β - glucosyl -2-S- benzyl -2, isothiobiuret (3g).**

<sup>1</sup>HNMR (CDCl<sub>3</sub>) : δ 7.34-7.15 (m, 10H, N-H(Ar.) and Ar-H); 6.98 (d, 1H, j=9Hz, NH(gal.)); 5.72 (t, 1H, j=9Hz, H<sub>1</sub>); 5.50 (d, 1H, j=3Hz, H<sub>4</sub>); 5.35-5.10 (m, 2H, H<sub>2</sub>, H<sub>3</sub>); 4.29-4.00 (m, 5H, H<sub>5,6,6'</sub> and Ar-CH<sub>2</sub>); 2.34 (m, 3H, Ar-CH<sub>3</sub>); 2.16-1.61 (m, 12H, 4COCH<sub>3</sub>). Anal. Calcd. for:- C<sub>30</sub>H<sub>35</sub>N<sub>3</sub>O<sub>9</sub>S<sub>2</sub>: C, 55.80; H, 5.42; N, 6.51; S, 9.92, found, C, 55.78; H, 5.48; N, 6.59; S, 9.99%.

**Table -1:- Physical data of 1-aryl – 5 – tetra – O – acetyl – β – glucosyl –2-S-benzyl- 2- isothiobiurets (3a-g) (Scheme-1) .**

Sr. No.	Reactant 1a-g (g)	Product (3a-g)	m. p. (°C)	Yield, g (%)	[α] <sub>D</sub> <sup>39</sup> (c, CHCl <sub>3</sub> )	R <sub>f</sub> (3:2, CCl <sub>4</sub> :EtOAc)
1	<b>1a</b> (1.21)	<b>3a</b>	174-176	2.1 (65.6)	122.2 <sup>0</sup> (0.9)	0.83
2	<b>1b</b> (1.38)	<b>3b</b>	116-118	2.5 (73.5)	-21.5 <sup>0</sup> (0.93)	0.84
3	<b>1c</b> (1.38)	<b>3c</b>	162-164	2.6 (76.4)	-84.2 <sup>0</sup> (0.95)	0.48
4	<b>1d</b> (1.38)	<b>3d</b>	158-160	2.2 (64.7)	85.1 <sup>0</sup> (0.94)	0.53
5	<b>1e</b> (1.28)	<b>3e</b>	136-138	2.1 (63.6)	52.0 <sup>0</sup> (0.96)	0.71
6	<b>1f</b> (1.28)	<b>3f</b>	150-152	2.4 (72.7)	-33.3 <sup>0</sup> (0.9)	0.77
7	<b>1g</b> (1.28)	<b>3g</b>	138-140	2.5 (75.7)	43.0 <sup>0</sup> (0.93)	0.70

**ACKNOWLEDGEMENT**

Authors are thankful to RSIC, CDRI Lucknow for providing the spectral data. Authors also thanks to Dr. S. G. Bhadange, Principal for encouragement and providing necessary facilities.

**REFERENCE**

1. **J. M. Garcia Fernandez; C. O. Mellet; (2000)** Chemistry and developments of N-thiocarbonyl carbohydrate derivatives: Sugar isothiocyanates, thioamides, thioureas, thiocarbamates, and their conjugates Adv. Carbohdr. Chem. Biochem. 55, 35 – 135.
2. **C.Prata; N.Mora; J.M.Lcombe, J.Maurizis; B. Pucci.( 1999)** Synthesis and molecular aggregation of new sugar bola-amphiphiles Carbohydr.Res., ,321, 4-14.
3. **M. M .De Oliveira;M. C. F. Linardi; M. R. P.Sampaio;(2006)** Effects of quinone derivatives on an experimental tumorJ.Pharm. Sci., 2006, 67, 562-563.
4. **Z. J. Witczak;(1986)** Monosaccharide Isothiocyanates and Thiocyanates: Synthesis, Chemistry, and Preparative Applications Adv. Carbohdr. Chem. Biochem.; 44, 91-145.
5. **Z. J. Witczak;(1986),** Desulfurization of glycosyl isothiocyanates with tributyltin hybriide Tetrahedron Lett; 27, 155-158
6. **J.Isac-Garcia; F. G.Calvo-Flores; F.Hernandez-Mateo; F.Santoyo- Gonzalez.( 2001);** Reactivity of 2-Deoxy-2-iodoglycosyl Isothiocyanates with O-, S-, and N-Nucleophiles. Synthesis of Glycopyranoso-Fused Thiazoles. Eur. J. Org. Chem., , 383-390
7. **Y.Gama; I.Shibuya; M.Shimizu;( 2002,)** Novel and Efficient Synthesis of 4-Dimethylamino-2- glycosylaminoquinazolines by Cyclodesulfurization of Glycosyl Thioureas with Dimethylcyanamide Chem. Pharm. Bull.; 50, 1517-1519.
8. **S. P. Deshmukh; B. N. Berad; M. G. Paranjpe;( . 1986)** J. Ind. Chem. Soc, LXIII, 315-316.
9. **G. V. Korpe; S. P. Deshmukh; (2002)** Synthesis of 1 tetra – O – benzoyl –  $\beta$  – D - Glucopyranosyl – 5 – aryl – 2 – S – benzyl – 2, 4 – isodithiobiurets and their antimicrobial activity.J. Ind. Chem. Soc. 79, 972-973.
10. **R.M. Silverstein; G. C. Bassler; T. C. Morill(1991)** “Spectrometric identification of Organic Compounds”, 5<sup>th</sup> Ed., John Wiley and Sons, INC, New York, N.Y.,.
11. **D.H. Williams and I. Fleming (1991)**“Spectrometric methods of Organic Chemistry,” 4<sup>th</sup> Ed., Tata-Mcgraw-Hill publication,
12. **J. R. Dyer,(1991)** “Applications of absorption spectroscopy of Organic Compounds,”8<sup>th</sup> Ed., Prentice Hall,
13. **J. M. J.Balnco; C. S. Barria; J. M. Bentio; C. O. Mellet; J.Fuentes; F. Santoyo-**

- Gonzalez; J. M. Garcia Fernandez; (1999,)** Synthesis, 11, 1904-1911.
- 14 **F.Sansone; E. Chierici; A. Casanti; R. Ungaro;(2003)**Thiourea link upper rim calix 4 arenes Org. Biomol. Chem.;; 1, 1802-1809.
- 15 **E. A. Werner;(1890)** J. Chem. Soc. Trans. , , 57, 283-304.
- 16 **E. Fischer; 1914,** Ber.47, 1377-1393.
- 17 **M. A. Saleh;(2002)** A convenient synthesis of novel nucleosides of 2-thioxo-5h-3,4-dihydropyrimido[5,4-b]indol-4-oneSulfur Lett. ;, 25, 235-245.