SYNTHESIS, CHARACTERIZATION AND BIOLOGICAL STUDIES OF NEWLY SYNTHESIZED 3-ARYL(NITROANILINE)-4-S-BENZYL-6-P-TOLYLIMINO-2-PHENYLIMINO-2,3-DIHYDRO-[1,3,5] THIADIAZINES[HYDROCHLORIDE]

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ABSTRACT

The progress achieved in the synthesis of heterocyclic compounds with biological potential is due to improvement of the methodological study of tested substance. Several five and six membered aromatic systems having three hetero atoms have been studied because of their interesting physiological properties. Serial of 3-Aryl(Nitroaniline)-4-S-benzyl-6-p-tolylimino-2-phenylimino-2,3-dihydro-[1,3,5] thiadiazine [Hydrochloride] has been synthesized by the interaction of 1-Aryl(Nitroaniline)-5-p-tolyl-2-S-benzyl-2,4-isodithiobiuretes with phenyl isocyanodichloride in refluxing chloroform medium. Initially evolution of hydrochloric gas to obtain 3-aryl (Nitroaniline)-4-S-benzyl-6-p-tolylimino-2,3-dihydro-1,3,5thiadiazines [hydrochloride]. Constitutions of synthesized compound have been delineated on the basis of chemical transformation, elemental determination, and IR, NMR and Mass spectral studies. These compounds were screened for their antibacterial and antifungal activities against–*Escherichia coli, Proteus vulgaris, Staphylococcus aureus, Salmonella typhimurium, Klebsiella pneumonie, Psudomonas aeruginosa, Aspergillus Niger* and *Candida albicance*. These compounds show appreciable activity towards these microorganisms.

Keywords: 2,4- isodithiobiuretes, phenyl isocyanodichlorides, -1, 3, 5-thiadiazines, Antimicrobial Activity

INTRODUCTION

Many compounds consisting of 5-membered heterocyclic rings represent important building blocks in organic and medicinal chemistry. In addition, they are interesting in their own right, due to their pharmacological properties¹⁻³. The progress achieved in the synthesis of heterocyclic compounds with biological potential is due to improvement of the methodological study of tested substance. Several five and six membered aromatic systems having three hetero atoms have been studied because of their interesting physiological properties. Their analogues have been noted to exert a wide range of clinical like antifungal⁵⁻⁶, antimalerial⁷, applications anti-HIV-1⁹, carbonic ahhydrase anticancer⁸, inhibitors¹⁰.

Thiadiazine and its derivatives are found as an important pharmacologically¹¹ and biologically active precursor in the field of heterocyclic chemistry. Some amino derivatives prove useful as herbicides, insecticides, fungicides, diuretics and ant diabetics. Organic thiocyanates ¹²⁻¹⁴ and sugar thiadiazines ¹⁵⁻¹⁷ also posses great potential as

carbonic anhydrase inhibitor, PET inhibitor, anti HIV agent, antitumor agent, psychotropic agent and used in treatment of breast cancer.

The heterocyclic compounds having 1, 3, 5-18-19 thiadiazine enhanced pharmaceutical medicinal, agricultural and industrial activities of the drugs and medicines. So the drugs or medicines containing thiadiazine nucleus are now used extensively in medical, biochemical and biotechnological faculties. The biological importance of the 1, 3, 5-thiadiazine derivatives is further emphasized by showing the presence of 1, 3, 5-thiadiazine ring in therapeutic agent

MATERIALS AND METHODS

All chemicals were research grade. Melting points determined are uncorrected. IR spectra were recorded in KBr on a FT-IR Perkin-Elmer RXI (4000-450cm⁻¹) spectrophotometer. ¹H NMR measurements were performed on a Bruker DRX-300 (300 MHz FT NMR) NMR spectrometer in CDCl₃ solution with TMS as internal reference. The Mass spectra were recorded on a THERMO Finnigan LCQ Advantage max ion trap Mass spectrometer. Optical rotation $[\alpha]_D^{31}$ measured on

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a Equip-Tronics Digital Polarimeter EQ-800 at 31^oC in CHCl₃. Thin layer chromatography (TLC) was performed on silica Gel G and spots were visualized by iodine vapour. The compounds describe in this paper were first time synthesized by the multistep reaction protocol

RESULTS AND DISCUSSION

Several 3-Arvl (Nitroaniline)-4-S-benzyl-6-ptolylimino-2-phenylimino-2,3-dihydro-[1,3,5] thiadiazine [Hydrochloride] (3a-d) have been synthesized by the interaction of 1-Aryl(Nitroaniline)-5-p-tolyl-2-S-benzyl-2,4isodithiobiurets (2a-d) and phenyl isocyanodichloride (1). То the chloroform suspension of Phenyl isothiocyanate chlorinated on chlorination assembly .This chlorinated

solution was added to 1-Aryl (Nitroaniline)-5-*p*-tolyl-2-*S*-benzyl-2,4-isodithiobiuret in chloroform medium . Then the reaction mixture was reflux for 3 hrs. and a sticky mass obtained as a residue was triturated several times with petroleum ether (60-80°C).. A product will separate out.

The IR spectra of products shows bands due to Ar-H, N-H, C=N, C-N, C-S stretching and ¹HNMR spectra of products distinctly displayed signals due to aromatic protons and aliphatic Protons. The Mass spectrum of product was also observed. The identities of these newly synthesized compounds have been established on the basis of usual chemical transformations and also IR, ¹H NMR and Mass spectral studies²⁰⁻²².

EXPERIMENTAL

1] Preparation of phenyl isocyanodichloride a] Preparation of phenyl isothiocyanate:^[22]

Place a 500ml conical flask in freezing mixture of ice and salt. Add to it 30 ml of conc. Ammonia solution and 15ml of pure carbon disulphide. Stir the mixture and run in 19 ml of aniline about 15 min. Stir for a further 30min and allow standing for another 30min. A heavy ppt of Ammonium phenyl dithiocarbamate separates. Filter it and dry it.

Transfer the salt to a 2lit R.B. flask. By 2-3 extractions with 100ml portion of distilled water. To this solution of 50gm of lead nitrate in distilled water with constant stirring. Lead sulphide ppt is observed. Steams distill the mixture into a receiver containing 10ml of $0.5ml H_2SO_4$ as long as organic material passes over. Separate the oil;

dry it over anhydrous Calcium Chloride or Magnesium Sulphate.

b] Preparation of phenyl isocyanodichloride

Through the chloroformic solution of phenyl isothiocyanate, chlorine gas was bubbled maintaining the temperature of the system below 10°C. After the addition of chlorine was completed, the yellow reaction mixture was diluted with 40-50ml petroleum ether (60-80°C). The solvent was then removed by distillation under vacuume. The whole operation was repeated several times with petroleum ether then phenyl isocyanodichloride was obtained as pale yellow oil.

2] Synthesis of 1-Aryl (Nitroaniline)-5-*p*-tolyl-2-*S*-benzyl-2,4-isodithiobiurets (2a-d)

a) Preparation of 1-Aryl (Nitroaniline)-S-benzyl isothiocarbamide

To the ethanolic suspension of Nitroaniline thiocarbamide was added benzyl chloride and the reaction mixture was reflux for 90 min. Afterward the reaction mixture was cooled and rendered basic with dil. ice cold NH_4OH and a sticky residue was obtained which on standing for 1 to 2 hrs. Solidifies. It was filtered and washes with petroleum ether.

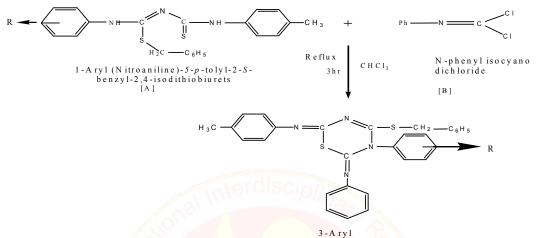
b) Preparation of 1-Aryl (Nitroaniline)-5-*p*-tolyl-2-*S*-benzyl-2,4-isodithiobiurets

Several 1-Aryl (Nitroaniline)-5-*p*-tolyl-2-*S*-benzyl-2,4-isodithiobiurets have been prepared by the interaction of 1-Aryl(Nitroaniline)-S-benzyl isothiocarbamide and *p*-tolyl isothiocyanate in benzene medium. To the benzene solution of 1-Aryl (Nitroaniline)-S-benzyl isothiocarbamide, ptolyl isothiocyanate was added .This reaction mixture was then reflux over a boiling water bath for 3 hrs. after completion of the reaction, solvent benzene was distilled off and the sticky mass obtained as residue was triturated several times with petroleum ether . A white product separated out crystallised from ethanol.

3] Synthesis of 3-Aryl (o-Nitroaniline)-4-Sbenzyl-6-p-tolylimino-2-phenylimino-2,3-

dihydro-[1,3,5] thiadiazine[Hydrochloride] 3-Aryl (o-Nitroaniline)-4-*S*-benzyl-6-*p*-tolylimino-2-phenylimino-2,3-dihydro-[1,3,5]

thiadiazine[Hydrochloride] was prepared by the interaction of 1-Aryl (o-Nitroaniline)-5-*p*-tolyl-2-*S*-benzyl-2,4-isodithiobiurets and phenyl isocyanodichloride in chloroform medium. A chloroform solution of Phenyl isocyanodichloride was mixed with the chloroform solution of 1-Aryl (o-Nitroaniline)-5-*p*-tolyl-2-*S*-benzyl-2,4isodithiobiuret. Then the reaction mixture was reflux on boiling water bath for 3 hr during which evolution of HCl was noticed. The progress of reaction was monitored by TLC. After completion of the reaction, the reaction mixture was brought to room temperature and the solvent removed under reduced pressure to obtain residue. This residue was triturated several times with petroleum ether (60-80°C) to afford a pale yellow solid.



(N itroaniline)-4-S-benzyl-6-p-tolylim ino-2-phenylim ino-2,3-dihydro-1,3,5-thiadiazine Hydrochloride

Where, R= (a) Phenyl, (b) *o*-Nitroaniline, (c) *m*-Nitroaniline, (d) *p*-Nitroaniline,

3a: IR (KBr): υ 3201 (Ar-H), 2877 (Ali-H), 1523 (C=N), 1450 (C-C), 1323 (C-N), 694 (C-S). H NMR (δ in ppm, CDCl₃): δ 7.63 -6.91 (19H, m, Ar. H), δ 4.27- 2.26 (5H, m, Ali. H) Mass (m/z): 490 (M⁺), 477, 387, 300, 91; Anal. Calcd for C₂₉H₂₄N₄S₂. C, 70.73; H, 4.87; N, 11.38; S, 13.00; Found: C, 70.70; H, 4.85; N, 11.40; S, 13.04.

On the basis of all above facts the product with m. p. 98°C was assigned the structure 3-Aryl(Phenyl)-4-S-benzyl-6-*p*-tolylimino-2-phenylimino 2,3dihydro- [1,3,5] thiadiazine [Hydrochloride]

When the reaction of phenyl isocyanodichloride was extended to several other 1-Aryl (Nitroaniline)-5-p-tolyl-2-S-benzyl-2,4-3-Aryl isodithiobiurets corresponding (Nitroaniline)-4-S-benzyl-6-p-tolylimino-2phenylimino 2,3-dihydro- [1,3,5] thiadiazine [Hydrochloride] has been synthesized. 3b: IR (KBr): 0 3082 (Ar-H), 2895 (Ali-H), 1504 (C=N), 1481 (C-C), 1303(C-N), 754 (C-S). H NMR (δ in ppm, CDCl₃): δ 8.08 -6.56 (18H, m, Ar. H), δ 4.42- 1.59 (5H, m, Ali. H) Mass (m/z): 537 (M⁺), 490, 462, 404, 390, 91; Anal. Calcd for C₂₉H₂₃N₅S₂O₂.; C, 64.80; H, 4.28; O, 5.95; N, 13.03; S, 11.91; Found: C, 64.85; H, 4.32; O, 5.90; N, 13.10; S, 11.89.

Compd	Yield	R _f	M.P.	Analysis (%): Found		
	%		⁰ C	(calcd)		
				Ν	S	
3a(Aniline)	75.00	0.55	98	11.40(11.38)	13.04(13.00)	
3b(o-Nitro-Phenyl)	68.00	0.60	130	13.00(13.03)	11.93(11.91)	
3c(m-Nitro-Phenyl)	78.00	0.48	135	13.05 (13.03)	11.98(11.91)	
3d(p-Nitro-Phenyl)	80.00	0.55	121	13.10 (13.03)	11.89 (11.91)	

Table -1: Physical data for characterization of compounds (3a-d)

C and H analysis was found satisfactory in all cases.

Antimicrobial activity²³:

All the compounds have been screened for both; antimicrobial and antifungal activity by using disc diffusion assay. For this, sterial filter paper disc (6 mm) impregnated with fixed doses of compounds was placed on pre-innoculated surface. The disc bearing plates were incubated at 37°C for 24 h. After incubation, zone diameters were measured.

The compounds were taken at a concentration or 1 mg/mL using dimethyl sulphoxide as a solvent. Amikacin (100 μ g/mL) was used as standard for antibacterial and fluconazole (100 μ g/mL) as a standard for antifungal activity. The compound were screened for antibacterial activity against *Eschrichia coli, Proteus vulgaris, Staphylococcus*

aureus, Salmonella typhi, Klebsiella pneumonie and Psudomonas aeruginosa in nutrient agar medium and for, antifungal activity against Aspergillus niger and Candida albicance in potato dextrose agar medium. It has been observed that all the compounds showed good activity against both; bacteria and fungi.

Compound	Ε.	S. aureus	P. vulgaris	P. aeruginosa	S. typhi	K. pneumonie	A. niger	C. albicance
	coli							
1(3a)	15	18	21	19	18	18	19	20
2(3b)	18	15	15	12	20	20	20	21
3(3c)	15	14	19	17	17	19	17	19
4(3d)	11	19	14	18	19	20	20	19
Amikacin	18	21	23	19	20	21		
Fluconazole							24	24

Zone of inhibition in mm. (15 or less) resistance, (16-20 mm) moderate and more than

CONCLUSION

In this research work, the characterizations of newly synthesized products were established on the basis of UV, IR, ¹H NMR, & Mass spectral studies. Various 3-Aryl (Nitroaniline)-4-*S*-benzyl-6-*p*-tolylimino-2-phenylimino 2,3-dihydro- [1,3,5] thiadiazine [Hydrochloride] were synthesized and yield of product ranged from 68-82%.

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REFERENCES

- Tetsuo Kawai, Mamoru Irie and Morihiko Sakaguchi, (1985) "Degradation of 2,4,6trialkyltetrahydro-1,3,5-thiadiazines during storage" J. Agric. Food Chem., 33 (3), 393-397.
- Yoshiyuki Kobayashi, Masao Shiozaki and Osamu Ando, (1995) "Synthesis: Development of a New o-Nitrobenzyl Photolabile Linker for Solid Phase" J. Org. Chem., 60 (8) 2570-2580
- A. A. El-Barbary, M. A. Sakran, A. M. El-Madani and C. Nielsen, (2005). "Synthesis and thermolysis of the 2,3-dihydro-1H-pyrole-2,3diones, pseudopericyclic reactions of formyl(N-phenylimidoyl)ketene" Journal of Heterocyclic Chemistry, 42 (5), 935-941.
- C. B. Vicentini, C. Romagnoli, E. Andreotti and D. Mares, (2004), "Development of New Fungicides against Magnaporthe grisea: Synthesis and Biological Activity of Pyrazolo[3,4-d][1,3]thiazine, Pyrazolo" Agri. Food Chem., 52(7), 2003-2009.

- 5. L. D. S. Yadav, A. Vaish and S. Sharma, (1994) "New fungitoxic fused-ring synthetics incorporating azoles and azines in different combinations" J. Agri. Food Chem., 42(3), 811-813.
- 6. L. C. March, G. S. Bajwa, J. Lee, K. Wasti and M. M. Joullie, (1976) "Synthesis and pesticidal activities of some substituted 1, 2, 4triazines" J. Med. Chem., 19(6), 845-848.
- A. K. Mansour, M. M. Eid and N. S. A. M. Khalil, (2003). "Synthesis and antimicrobial activity of chromone-linked 2-pyridone fused with 1,2,4-triazoles, 1,2,4-triazines and 1,2,4triazepines ring systems" Nucleosides, Nucleotides and Nucleic Acids, 22(1), 21-44.
- Z. Brzozowski, F. Saczewski, J. Slawinski, T. Sanchez and N. Eamati, (2009) "Synthesis and QSAR Study of Novel 6-Chloro-3-(2-Arylmethylene-1-methylhydrazino)-1,4,2-benzodithiazine 1,1-Dioxide Derivatives with Anticancer Activity" Eur. J. Med. Chem., 44, 190-196.

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- A. R. Katritzky, K. C. Caster, T. H. Maren, C. V. Conroy and A. Bar-Llan, (1987) "Macrolides with gastrointestinal motor stimulating activity" J. Med. Chem., 30 (11), 2058-2062.
- C. B. Vicentini, D. Mares, A. Tartari, M. Manfrini and G. Forlani, (2004) "Development of New Fungicides against Magnaporthe grisea: Synthesis and Biological Activity of Pyrazolo[3,4-d][1,3]thiazine, Pyrazolo" J. Agri. Food Chem., 52(7), 1898-1906.
- K. Tsujihara, M. Ozeki, T. Morikawa, M. Kawamori, Y. Akaike and Y. Arai, (1982) "Synthesis and antitumor activity of disaccharide derivatives of 3,3-disubstituted 1-(2-chloroethyl)-1-nitrosoureas" J. Med. Chem., 25, 441.
- 12. T. K. Lindhorst and C. Kieburg, (1995) "A Comprehensive Review of the Synthetic Literature "Synthesis, 10, 1228.
- Ortizmellet, J. Balanco , L. Jose, G. Fernandez and M. Jose, Fuentes, (1993).
 "Building Blocks for Glycopeptide Synthesis. Disaccharide Glycosyl Isothiocyanates" J. Carbohydr. Chem., 12, 487
- 14. R. D. Ghuge and S. P. Deshmukh, 1391 (2011). "SYNTHESIS OF N-MALTOSYLATED THIOCARBAMIDES, Int. J. Chem. Sci., 9(3),
- P. R. Mahalle, G.V. Korpe and S. P. Deshmukh , (2008) "New N-galactosides: Synthesis of N-galactosylated thiocarbamides, benzothiazolyl thiocarbamides and thiocarbamates" Ind.J.Heterocyclic Chem., 18, 161

- M. T. Sangole and S. P. Deshmukh , (2011). "Synthesis of 2-Phenylimino-3-aryl-4-Sbenzyl-6-hepta-O-benzoyl-β-D-maltosylimino -2,3-dihydro-1,3,5-thiadiazines (Hydrochloride)" J. Ind. Chem. Soc., 88, 739.
- 17. D.T Tayade., Ph.D. Thesis, Amravati Univ, 1996.
- M. G. Paranjape (1966), "Synthesis and Structural Elucidation of A New Series of 1-Formamidino-(N-Substitutedthioamido)-5-Substituted-2-Thio-4-IminoBiuret" J. Indian Chem. Soc., 42-45.
- R. M. Silverstein, G. C. Bassler and T. C. Morrill, Spectrometric Identification of Organic Compounds", 5th ed., John Wiley and Sons, Inc, New York, (1991).
- 20. N.B. Colthup, L.H. Daly and S.E. Wibereley, Introduction to Infrared and Raman spectroscopy "Academic Press, New York, (1964).
- D.H.Williams and L.Fleming, Spectroscopic Meyhods in Organic Chemistry", 4th ed., Tata-McGraw- Hill, (1988).
- 22. Furniss, A.J. Hannaford, W.G. Smith, A.R. Tatchell , "Vogel text book of practical organic chemistry", FLBS, 4th ed.; p (a) 456 (b)
- J.G. Collee and W Marr Laboratory control of anti-microbial therapy. In: college J G, Duguid J P, Fraser A G, Marmion B P (eds) Mackie and Mc Cartney Practical Medical Microbiology. 14th edition. Edinburgh Churchill Livingeston, (1996), 114-115